

"Through an older eye"

Antique cameras, historical processes and an old(ish) photographer

Simon Williams 2024 oldparson.art











Printing out technologies

- 1. Salt print (Silver)
- 2. Albumen (Silver)
- 3. Cyanotype (Iron)
- 4. Van-Dyke Brown (Argyrotypes) (Silver & Iron)
- 5. Silver-gelatine
- 6. Cyanotype-gelatine

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by Hill & Adamson, a salt print from a <u>calotype</u> paper negative, c. 1844





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Louis Désiré Blanquart-Evrard in 1869, albumen print, by himself





Silver-based Chemistry

- Silver-Halide salts are light sensitive - Silver Chloride AgCl (react Silver Nitrate with Sodium Chloride
- Silver Bromide AgBr
- Silver Iodide Agl

Light energy 'excites' the atomic structure of the Silver-Halide - invisible unless intense, & or, sustained light applied.

Developer latches onto the excited areas and forms silver metal - and "amplifies" the effect making the latent (hidden) image visible.

- "Fix" removes unused silver salts from the paper/film (Sir John Herschel)



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Early cyanotype images



1842 - Lady with Harp print of an engraving

Sir John Herschel

1840s - Photogram of seaweed

Anna Atkins

Cyanotyope Chemistry

potassium ferricyanide then dried.

ferric iron (Fe III) of the citrate into divalent ferrous iron (Fe II).

This reacts with the potassium ferricyanide forming insoluble divalent iron.

light received.

- Paper is impregnated with a mix of dissolved ferric ammonium citrate and
- When the paper is exposed to UV light, a photo-reaction turns the trivalent
- ferroferricyanide Fe₄(Fe(CN)₆)₃ (Prussian blue or Turnbull's blue) with the
- Unaltered ammonium ferric citrate and potassium ferricyanide, from the shaded areas are then washed away in inverse proportion to the amount of

Cyanotype -straight & toned with tannin (coffee, tea etc)









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Hands-on with Cyanotype

- **1.** Initial paper
- 2. Mix equal volumes of solutions A+B to
- make green cyanotype sensitiser
- **3.**Coat paper evenly with sensitiser don't
- overwork & roughen paper.
- 4.Dry paper
- **5.**Load contact frame:
- Neg onto glass, then protective film then sensitised paper
- 6. Make exposure (15 mins probably until get grey areas)



- 7. Make exposure (15 mins probably -
- until get grey areas)
- 8. Rinse in acidic water
- 9. Peroxide spray
- **10.** Optional bleach & wash
- **11.** Optional tone & wash
- 12. Hang to dry (will darken)

The components of the Herschel cyanotype formula are pretty benign: ferric ammonium citrate is not poisonous (it is taken medicinally, and used (E381) in a celebrated Scottish soft drink 'Irn-Bru'); its components, ferric ion and citrate ion, are widely naturally-occurring. Ferricyanide ion would quickly reduce in the environment to ferrocyanide, which is also non-toxic; it is a permitted food additive (E536) used as an anti-caking agent in table salt. ((c)Mike Ware 2017)

Safety

Cyanotype supplies



Free 377-Page treatise on all things Cyanotype https://www.mikeware.co.uk/downloads/Cyanomicon_II.pdf



 \checkmark

This item: BOSITE UV LED Black Light,395nm 30W Ultraviolet Floodlight IP66 (85V-265V AC)...

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Photography can be

Journalism Social media Historical record Aide-memoire (Holiday snaps) Hunting (wildlife) Text book illustration Advertising Science Art

Photography as "Art"

Writing can be: prose - conveying information poetry - communicating more than the literal meaning of each word.

Photography can be Art, like poetry - conveying more than the literal understanding of the image presented

Pictoral Sm

was as much "art" as was painting and drawing by hand.

Composition was guided by the traditions of drawing/painting

Soft focus lenses and textured papers

output of a machine.

Printing processes such as Gumoil and Photogravure produced effects that approximated the look of drawings, etchings, and oil paintings.

- A school of photography arising from the desire to show photography
- Printing techniques were used that showed traces of the hand of the photographer in the process - the image was not merely a technical



